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The Rachford-Rice function $f\{Y\}$ can be used to find phase conditions at equilibrium for flash separations in multicomponent systems $c\{4\} = 3$ $Z_i(1 - K_i) = 0$ $1 + Y(K_i - 1)$ $i=1$ Where Y is the fraction of feed that vaporizes ($Y = \text{vapor/feed}$), z_i is the mole fraction of the i th component in the feed, and K_i is the phase equilibrium ratio, commonly simply referred to as the K -value ($K_i = y_i/x_i$).

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